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Updated: 10/18/2017

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UNIT 4 PRACTICE TEST

Honour Chemistry: Unit 4 Practice Test: Thermochemistry and Nuclear Chemistry Part A: Multiple Choice and Numerical Response (1 mark each for Multiple Choice) (2 marks each for Numerical Response) $E_{\text{bind}} = \sum mc^2$
 c = speed of light (3.00 10⁸ m s⁻¹) 1 eV = 1.69 10⁻¹⁹ J (1 kg = 6.022 10²⁶ amu , or 1 g = 6.022 10²³ amu = 1 mole amu) 1 MeV = 1.69 10⁻¹⁶ J (1 Neutron = 1.008665 amu) (1 Proton = 1.007825 amu)

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listening testpage 1 As you listen to the tape, fill in the missing information for questions 110. The conversation will be played twice. An example has been done for you. MIKE: Gday, Jane. JANE: Gday, Mike. MIKE: Im heading off to Paddys Market on Sunday. JANE: What are you going to do there? MIKE: Well, I want to buy some fruit, oh, and a pair of sunglasses too. JANE: Actually, I wouldnt mind joining...

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MICROSOFT WORD - UNIT 6 PRACTICE TEST

Unit 6 Practice Test Energy Show all your work. NO WORK = NO CREDIT 1. Find the heat (in calories) involved when raising 150 grams of water from 20 oC to 70 oC. Was the heat gained or lost by the water? 2. What mass of iron can be colled from 80 oC to 50 oC if 1379.4 J is given off? The specific heat of iron is 0.45J g*C 3. When 435 joules of heat energy is added to 3.40 grams of olive oil at 21 o...

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